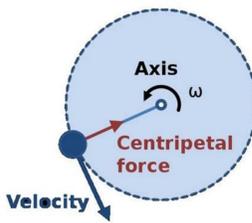
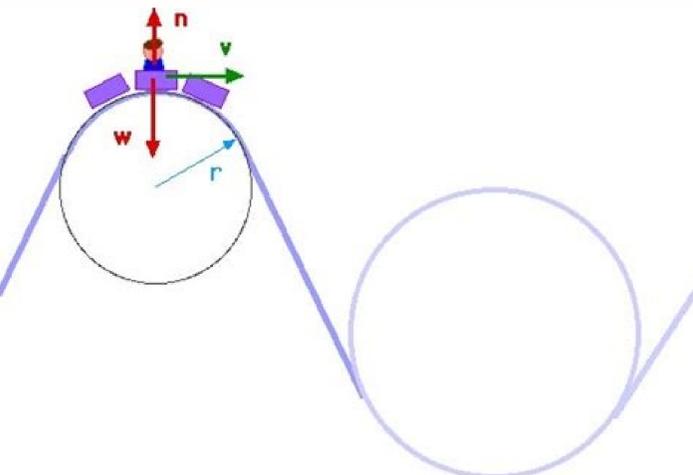
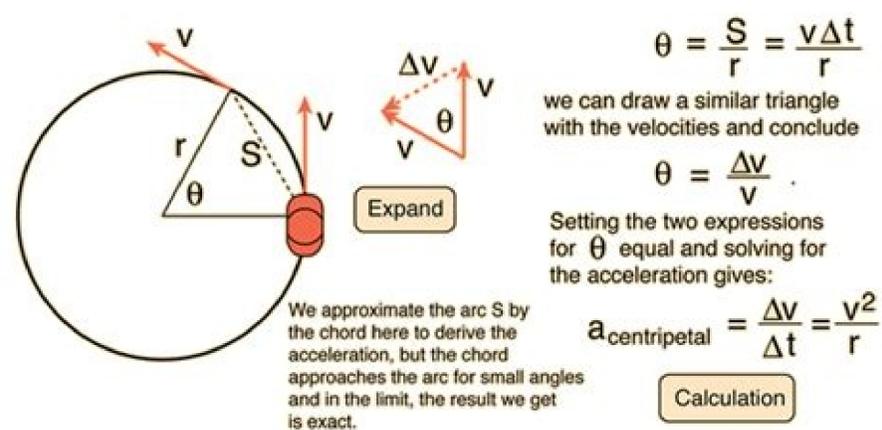


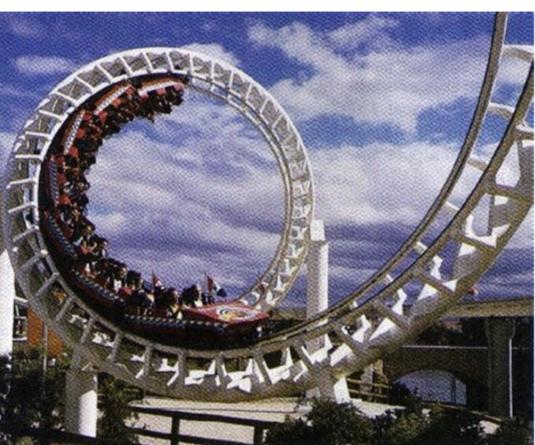
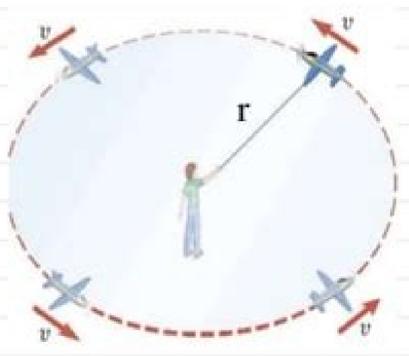
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## Uniform Circular Motion

$v$  is constant  
 $\vec{v}$  is constantly changing



In order to continue enjoying our site, we ask that you confirm your identity as a human. Thank you very much for your cooperation. Course Coverage Update Colleges agree that Units 8-10 can be removed from AP Physics 1 since they are covered in AP Physics 2; accordingly, Units 8-10 are no longer tested in AP Physics 1. As of 2021, AP Physics 1 Exams focus exclusively on content covered in Units 1-7. Learn about the foundational principles of physics as you explore Newtonian mechanics; work, energy, and power; mechanical waves and sound; and introductory, simple circuits. You'll do hands-on laboratory work to investigate phenomena. Note: Save your lab notebooks and reports; colleges may ask to see them before granting you credit. Interpreting and describing representations and models Using mathematics to solve science problems Formulating a scientific question or hypothesis Designing an experiment to answer a scientific question or to test a hypothesis Analyzing data and evaluating evidence Working with scientific explanations and theories Making connections A first-semester introductory college course in algebra-based physics You should have completed a geometry course and be concurrently taking Algebra II or an equivalent course. The course content outlined below is organized into commonly taught units of study that provide one possible sequence for the course. Your teacher may choose to organize the course content differently based on local priorities and preferences. Course Content Topics may include: Position, velocity, and acceleration Representations of motion Topics may include: Systems The gravitational field Contact forces Newton's First Law Newton's Third Law and free-body diagrams Newton's Second Law Applications of Newton's Second Law Topics may include: Vector fields Fundamental forces Gravitational and electric forces Gravitational field/acceleration due to gravity on different planets Inertial vs. gravitational mass Centripetal acceleration vs. centripetal force Free-body diagrams for objects in uniform circular motion Topics may include: Open and closed systems: Energy Work and mechanical energy Conservation of energy, the work-energy principle, and power Topics may include: Momentum and impulse Representations of changes in momentum Open and closed systems: momentum Conservation of linear momentum Topics may include: Period of simple harmonic oscillators Energy of a simple harmonic oscillator Topics may include: Rotational kinematics Torque and angular acceleration Angular momentum and torque Conservation of angular momentum In order to continue enjoying our site, we ask that you confirm your identity as a human. Thank you very much for your cooperation. This fMRI scan shows an increased level of energy consumption in the vision center of the brain. Here, the patient was being asked to recognize faces. Image credit: NIH via Wikimedia Commons All bodily functions, from thinking to lifting weights, require energy. The many small muscle actions accompanying all quiet activity, from sleeping to head scratching, ultimately become thermal energy, as do less visible muscle actions by the heart, lungs, and digestive tract. The rate at which the body uses food energy to sustain life and to do different activities is called the metabolic rate. The total energy conversion rate of a person at rest is called the basal metabolic rate (BMR) and is divided among various systems in the body, as shown the following table: Basal Metabolic Rates (BMR) Organ Power consumed at rest (W) Oxygen consumption (mL/min) Percent of BMR Liver & spleen 23 67 27 Brain 16 47 19 Skeletal muscle 15 45 18 Kidney 9 26 10 Heart 6 17 7 Other 16 48 19 Totals 85 W 250 mL/min 100% The largest fraction of energy goes to the liver and spleen, with the brain coming next. About 75% of the calories burned in a day go into these basic functions. A full 25% of all basal metabolic energy consumed by the body is used to maintain electrical potentials in all living cells. (Nerve cells use this electrical potential in nerve impulses.) This bioelectrical energy ultimately becomes mostly thermal energy, but some is utilized to power chemical processes such as in the kidneys and liver, and in fat production. The BMR is a function of age, gender, total body weight, and amount of muscle mass (which burns more calories than body fat). Athletes have a greater BMR due to this last factor. Of course, during vigorous exercise, the energy consumption of the skeletal muscles and heart increase markedly. The following diagram summarizes the basic energetic functioning in the human body. The most basic functions of the human body mapped to the main concepts covered in this textbook. (Chemical potential energy is actually a form of electric potential energy, but we will not specifically discuss electric potential energy in this textbook so we have separated the two.) Heat The body is capable of storing chemical potential energy and thermal energy internally. Remembering that thermal energy is just the kinetic energy of atoms and molecules, we recognize that these two types of energy are stored microscopically and internal to the body. Therefore, we often lump these two types of microscopic energy into the (). When an object is warmer than its surroundings then will be transferred from object to surroundings, but if the object is cooler than its surroundings then thermal energy will transferred into the object from its surroundings. The amount of thermal energy exchanged due to temperature differences is often called (). When heat is transferred out of the body to the environment, we say call this, as indicated in the previous figure. We will learn more about how temperature and heat transfer are related in the next unit. Energy Conservation The Principle of Conservation of Energy states that energy cannot be created or destroyed. Therefore, if the body does useful work to transfer to its surroundings (), or transfer to the environment as, then that energy must have come out of the body's. We observe this throughout nature as the First Law of Thermodynamics: (1) Heat Engines Your body uses chemical potential energy stored internally to do, and that process also generates thermal energy, which you release as. The internal combustion engines that power most cars operate in similar fashion by converting chemical potential energy in fuel to thermal energy via, then converting some of the thermal energy into and dumping some into. Your body is capable of releasing the chemical potential energy in your food without combustion, which is good, because you are not capable of using your from your to do. Machines that can use to do work, such as a combustion engine, are known as. Heat engines are still governed by the First Law of Thermodynamics, so any must have been thermal energy that was not used to do work. The thermal energy input that can be used to do work rather than wasted as determines the of the heat engine. The of the human body in converting chemical potential energy into is known as the of the body. We often calculate the body's mechanical efficiency, as a percentage: (2) The mechanical efficiency of the body is limited because energy used for metabolic processes cannot be used to do useful work. Additional generated during the chemical reactions that power muscle contractions along with in joints and other tissues reduces the efficiency of humans even further.. "Alas, our bodies are not 100 % efficient at converting food energy into mechanical output. But at about 25 % efficiency, we're surprisingly good considering that most cars are around 20 %, and that an Iowa cornfield is only about 1.5 % efficient at converting incoming sunlight into chemical [potential energy] storage." For an excellent discussion of human and comparisons with other machines and fuel sources, see MPG of a Human by Tom Murphy, the source of the previous quote. Assuming a 20% in climbing stairs, how much does your decrease when a 65 kg person climbs a 15 m high flight of stairs? How much does the person transfer to the environment as? First, let's calculate the change in gravitational potential energy: The person did in converting chemical potential energy in their body to mechanical energy, specifically gravitational potential energy. However, they are only 20% efficient, which means that only 1/5 of the chemical potential energy they use goes into doing useful work. Therefore the change in chemical potential energy must have been 5x greater than the mechanical work output. The chemical potential energy used came out of the person's so: We can use the First Law of Thermodynamics to find the thermal energy exhausted by the person: (3) Rearranging for: We find that the heat is negative, which makes sense because the person exhausts thermal energy out of the body and into to the environment while climbing the stairs. Alternatively, we could have known right off that the must be 4/5 of the total loss of, because only 1/5 went into doing useful. So the exhaust heat should be: For historical reasons we often measure and in units of (cal) instead of. There are 4.184 Joules per calorie. We measure chemical potential energy stored in food with units of 1000 calories, or kilocalories (kcal) and we sometimes write kilocalories as Calories (Cal) with with capital C instead of a lowercase c. For example, a bagel with 350 Cal has 350 kcal, or 350,000 cal. Converting to Joules, that would be in the bagel. What fraction of a bagel would you need to eat in order to make up for the 47,775 J loss of internal energy (as chemical potential energy) that we calculated in the previous everyday example about climbing stairs? There are 1,464,400 J/bagel Therefore we need to eat: A pulse oximeter is an apparatus that measures the amount of oxygen in blood. Oxymeters can be used to determine a person's metabolic rate, which is the rate at which food energy is converted to another form. Such measurements can indicate the level of athletic conditioning as well as certain medical problems. (credit: UusiAjaja, Wikimedia Commons) The digestive process is basically one of oxidizing food, so energy consumption is directly proportional to oxygen consumption. Therefore, we can determine the the actual energy consumed during different activities by measuring oxygen use. The following table shows the oxygen and corresponding energy consumption rates for various activities. Energy and Oxygen Consumption Rates for an average 76 kg male Activity Energy consumption in watts Oxygen consumption in liters O<sub>2</sub>/min Sleeping 83 0.24 Sitting at rest 120 0.34 Standing relaxed 125 0.36 Sitting in class 210 0.60 Walking (5 km/h) 280 0.80 Cycling (13-18 km/h) 400 1.14 Shivering 425 1.21 Playing tennis 440 1.26 Swimming breaststroke 475 1.36 Ice skating (14.5 km/h) 545 1.56 Climbing stairs (116/min) 685 1.96 Cycling (21 km/h) 700 2.00 Running cross-country 740 2.12 Playing basketball 800 2.28 Cycling, professional racer 1855 5.30 Sprinting 2415 6.90 In the previous examples we assumed stated that our mechanical efficiency when climbing stairs was 20%. Let's use the data in the above table to verify that assumption. The data in the table is for a 76 kg person climbing 116 stairs per minute. Let's calculate the rate at which that person did mechanical work while climbing stairs and compare that rate at which they used up internal energy (originally from food). The minimum standard step height in the US is 6.0 inches (0.15 m) then the gravitational potential energy of a 76 kg person will increase by 130 J with each step, as calculated below: When climbing 116 stairs per minute the rate of energy use, or power, will then be: According to our data table the body uses 685 W to climb stairs at this rate. Let's calculate the efficiency: As a percentage, this person is 32% mechanically efficient when climbing stairs. We may have underestimated in the previous examples when we assumed a 20% efficiency for stair climbing. We often talk about "burning" calories in order to lose weight, but what does that really mean scientifically? First, we really mean lose because that is the measure of how much stuff is in our bodies and weight depends on where you are (it's different on the moon). Second, our bodies can't just interchange and — they aren't the same physical quantity and don't even have the same units. So how do we actually lose mass by exercising? We don't actually shed the atoms and molecules that make up body tissues like fat by "burning" them. Instead, we break down the fat molecules into smaller molecules and then break bonds within those molecules to release chemical potential energy, which we eventually convert to and. The atoms and smaller molecules that resulting from breaking the bonds combine to form carbon dioxide and water vapor (CO<sub>2</sub> and H<sub>2</sub>O) and we breath them out. We also excrete a bit as H<sub>2</sub>O in sweat and urine. The process is similar to burning wood in campfire — in the end you have much less of ash than you did original wood. Where did the rest of the mass go? Into the air as CO<sub>2</sub> and H<sub>2</sub>O. The same is true for the fuel burned by your car. For more on this concept see the first video below. The really amazing fact is that your body completes this chemical process without the excessive temperatures associated with burning wood or fuel, which would damage your tissues. The body's trick is to use enzymes, which are highly specialized molecules that act as catalysts to improve the speed and of chemical reactions, as described and animated in the beginning of the second video below. Similar to the body, the efficiency of any energetic process can be described as the amount of energy converted from the input form to the desired form divided by the original input amount. The following chart outlines the efficiencies of various systems at converting energy various forms. The chart does not account for the cost, hazard risk, or environmental impact associated with the required fuel, construction, maintenance, and by-products of each system. The Efficiency of the Human Body Compared to Other Systems

System Input Energy Form Desired Output Form Max Efficiency Human Body Mechanical 25 % Coal/Oil/Gas Fired Stream Turbine Power Plants Mechanical 47% Combined Cycle Gas Power Plants Electrical 58 % Biomass/Biogas Kinetic Electrical 40% Nuclear Kinetic Electrical 36% Solar-Photovoltaic Power Plant Sunlight (Electromagnetic) Electrical 15% Solar-Thermal Power Plant Sunlight (Electromagnetic) Electrical 23% Hydroelectric and Tidal Power Plants Gravitational Potential Electrical 90%+ Check out the energy systems tab in this simulation to visualize different energy conversion systems energy stored in the chemical bonds of a substance the total of a systems thermal energy and chemical potential energy, the total energy stored microscopically within the system energy stored in the microscopic motion of atoms and molecules (microscopic kinetic energy) An amount of thermal energy transferred due to a difference in temperature. heat transferred to the environment rather than being used to do useful work Energy cannot be created or destroyed, only transferred from one type to another and/or from one object to another the sum of potential and kinetic energy the change in internal energy of a system is equal to the heat added to the system minus the work done by the system A quantity representing the effect of applying a force to an object or system while it moves some distance. the process of burning, the rapid chemical combination of a substance with oxygen, involving the production of heat and light work done on the external environment, such as moving objects, as opposed to work done internally, such as pumping blood devices for converting thermal energy to useful work and exhaust heat ratio of useful work performed to total energy expended the effectiveness of a machine in transforming the energy input to the device to mechanical energy output a force that acts on surfaces in opposition to sliding motion between the surfaces potential energy stored in objects based on their relative position within a gravitational field unit of energy equivalent to 4.184 Joules International standard (SI) unit of Energy a measurement of the amount of matter in an object made by determining its resistance to changes in motion (inertial mass) or the force of gravity applied to it by another known mass from a known distance (gravitational mass). The gravitational mass and an inertial mass appear equal. A quantity representing the capacity of an object or system to do work.

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Centripetal force is perpendicular to tangential velocity and causes uniform circular motion. The larger the centripetal force  $F_c$ , the smaller is the radius of curvature  $r$  and the sharper is the curve. The lower curve has the same velocity  $v$ , but a larger centripetal force  $F_c$  produces a smaller radius  $r$  . In the section on uniform circular motion, ... the relationship between force and acceleration, and the nature of force pairs between objects. In addition, the High School Physics Laboratory Manual addresses content in this section in the lab titled: Circular and Rotational Motion, as well as the following standards: (4) ... 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